

BIOGENETIC-TYPE CYCLIZATION REACTIONS OF GERMACRONES IN THIOPHENOL

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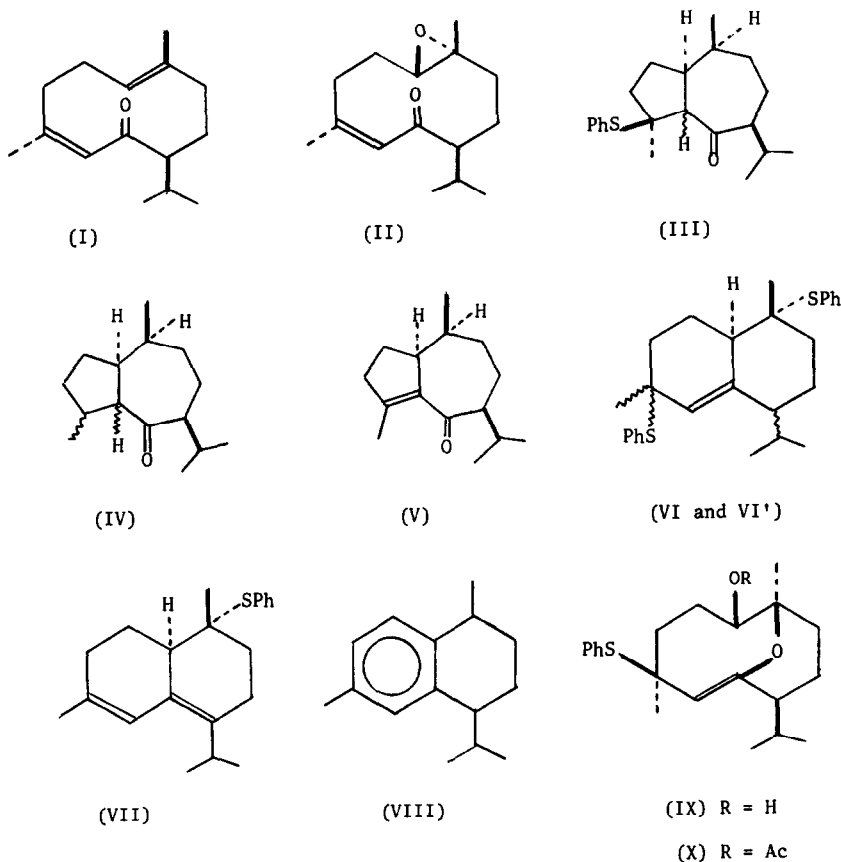
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Previously, we reported the biogenetic-type reactions of the germacrones.¹ We further examined the acid-catalysed cyclizations of the germacrones (I and II) in order to control the direction of the intramolecular cyclization reactions depending on conformations of the reactive intermediates of the germacrones. In fact, cyclization products varied with different kinds of acid (HCOOH, AcOH and H₂SO₄). Particularly, the formation of a guaiane-type compound (III) from I should be noted, as mentioned below.

The acid-catalysed cyclization of the germacrone (I) with 80% aq.AcOH has been known to afford the cadinane-type compound.¹ On the other hand, when treated with AcOH in PhSH (room temp., 24hr) instead of 80% aq.AcOH, I was converted into a guaiane-type compound (III) in 50% yield, the structure of which was confirmed by its spectral data coupled with the following chemical evidences [m.p. 90-91°; C₂₁H₃₀O₅; m/e 330(M⁺) and 221; ν_{\max} (KBr) 1700cm⁻¹; δ (C₆D₆) 0.87(6H, d, J= 6.7Hz), 1.06(3H, d, J= 6.3Hz), 1.21(3H, s), 2.75(1H, d, J= 8.5Hz) and 7.00-7.28ppm(5H, m)]. Its NMR spectrum indicates the presence of a secondary methyl group in addition to the isopropyl group. Furthermore, the doublet at δ 2.75ppm can be assigned to a methine proton at C₅-position, which is coupled to an adjacent methine proton at C₁-position (J= 8.5Hz). Further treatment of III with Raney Ni in abs.EtOH under N₂ (under reflux, 6hr) gave a desulphurization product (IV) in 86% yield; m.p. 55-57°; C₁₅H₂₆O; m/e 222(M⁺); ν_{\max} (KBr) 1700cm⁻¹; δ (CDCl₃) 0.83(3H, d, J= 6.7Hz), 0.87(3H, d, J= 6.5Hz), 0.92(3H, d, J= 6.5Hz), 1.13(3H, d, J= 6.3Hz) and 2.86ppm(1H, q, J= 7.8 and 4.8Hz); ORD curve ($[\phi]_{321}^P +14^\circ \times 10^2$; $[\phi]_{276}^T -39^\circ \times 10^2$; A= +53). In particular, the appearance of a methine quartet at δ 2.86ppm indicates that the PhS group

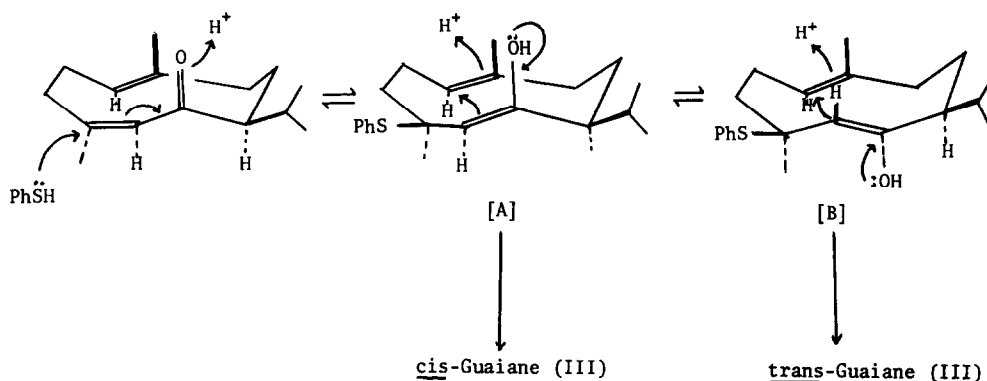
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in III must be located at C₄-position. This was further confirmed by action of MeI in acetone (under reflux, 8hr) on III, leading to the formation of a fully substituted α - β -unsaturated ketone (V) in 66% yield: C₁₅H₂₄O; m/e 220(M⁺); ν_{\max} (film) 1670 and 1610cm⁻¹; λ_{\max} (MeOH) 256nm (ϵ , 8270). From the above facts, the structure of the cyclization product should be represented by III except for its stereostructure. The relative configurations at the newly formed asymmetric carbon atoms in III remain unsettled. However, thiophenol



attacks the conjugated double bond from a β -side in the known conformation of I,¹ and then the cyclization reaction must take place in a concerted manner leading to the formation of III, as described below. In this case, the cis-guaiane (III) may be formed from a possible intermediate [A]. On the other hand, if conformational change of [A] to

another intermediate [B] takes place, the trans-guaiane (III) must be formed.² Unfortunately, any complete evidences on the configuration at C₅-position have not yet been obtained.³



In the case of 100% HCOOH in PhSH-benzene (1 : 1) (room temp., 15hr) instead of AcOH, a mixture of four cyclization products (III, VI and VI', and VII) were obtained in 15, 12 and 5% yields, respectively. The inseparable mixture (VI and VI') can be easily converted into the conjugated diene (VII) as follows. When treated with MeI in acetone under N₂ (under reflux, 4hr), the mixture afforded the diene (VII) and calamenene (VIII), in 45 and 29% yields, respectively. On the other hand, action of conc.H₂SO₄ (0°, 30min) on VII gave the mixture (VI and VI') in 75% yield. The structure of the diene (VII) was confirmed by its spectral data: C₂₁H₂₈S (m/e 312(M⁺)); δ(CCl₄) 0.93(6H, d, J= 7.0Hz), 1.09(3H, s), 1.76(3H, br.s), 2.99(1H, m), 6.13(1H, br.s) and 7.30ppm(5H, m). Finally, the acid-catalysed cyclization reaction of I was effected with conc.H₂SO₄ in PhSH(0°, 30min) leading to the formation of the cadinane-type compounds (VI and VI', and VII) in 67 and 4% yields, respectively. In this case, any guaiane-type compound has not been isolated. Probably, these cadinanes are formed according to the same concerted mechanism as previously reported.¹

The epoxy-germacrone (II) having the same conformation as that of I¹ was also treated with 100% HCOOH in PhSH (room temp., 40min) to give a hydroxy-olefin (IX) in 66% yield (C₂₁H₃₀O₂S; m/e 316(M⁺- 110); ν_{max} (film) 3420br., 3050, 1665, 1580, 1040, 745 and 690cm⁻¹; δ(CDCl₃) 0.86(3H, d, J= 6.0Hz), 0.88(3H, d, J= 6.0Hz), 1.21(3H, s), 1.35(3H, s), 1.49(1H, s; OH), 4.25(1H, t, J= 6.0Hz), 4.75(1H, s) and 7.30ppm(5H, complex)), which was converted into the corresponding acetate (X), in 86% yield, on acetylation with Ac₂O -

pyridine (0°, overnight). In the NMR spectrum of X, the triplet at δ 4.25ppm in IX was observed at δ 5.57ppm(1H, br.t, $J \sim 6.8$ Hz). In the above reaction, any selinane-type compound has not been obtained.¹

All compounds gave satisfactory physical data, and their structures were confirmed by IR, UV, NMR and mass spectrometric data.

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REFERENCES

1. M. Iguchi, M. Niwa and S. Yamamura, Chem. Commun., 974 (1971); *ibid.*, 689 (1972).
2. J. A. Marshall, W. F. Huffman and J. A. Ruth, *J. Amer. Chem. Soc.*, 94, 4691 (1972).
3. Although the base-catalysed equilibrium experiment on IV has revealed that this compound was less stable than the newly formed isomer [C₁₅H₂₆O; m/e 222(M⁺); ν_{\max} (film) 1695cm⁻¹; δ (CDCl₃) 0.84-1.06ppm(12H, complex); ORD curve ($[\phi]_{314}^T$ -46° x 10²; $[\phi]_{272}^P$ +57° x 10²; A= -103], further studies on this point have been required.